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Acta Cryst. (1995). C51, 2698-2700

(Z)-2,3-Bis(3,4-dimethoxyphenyl)propenoic Acid

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(Received 31 January 1995; accepted 5 July 1995)

Abstract

The crystal structure determination demonstrates unambiguously that the title compound, $C_{19}H_{20}O_6$, has the Z configuration. The compound was synthesized starting from 2,3-bis(3,4-dimethoxyphenyl)-2-hydroxypropanoic acid, which was in turn obtained by an alkaline treatment of *trans*-1,3-bis(3,4-dimethoxyphenyl)-2,3-epoxy-1-propanone. ¹H NMR data for the title compound and its methyl ester are reported.

Comment

 α -Arylcinnamic acids are of interest as intermediates in the syntheses of stilbenes. Gierer, Lenic, Norén & Szabo-Lin (1974) prepared a series of lignin-related stilbenes using α -arylcinnamic acids as starting materials. Such compounds (and their esters) can also be used as starting materials for the preparation of α -arylcinnamyl alcohols. Certain compounds of this latter type are intermediates in the acid-catalysed reactions of lignin models of the 1,2-diaryl-1,3-propanediol type (Li & Lundquist, 1995). The X-ray work presented in this study establishes the structure and stereochemistry of such an α -arylcinnamic acid, namely, (Z)-2,3-bis(3,4-dimethoxyphenyl)propenoic acid, (1). Since the synthesis of α -arylcinnamyl alcohols from α -arylcinnamic acids usually proceeds *via* the ester derivatives, we also prepared the methyl ester of (1), *i.e.* (2), and examined it by X-ray crystallographic methods. Attempts to obtain good crystals of (2) failed (the crystals often suddenly cracked, a possible explanation being internal strain). However, a structure analysis was performed and it could be concluded that (2) had the expected Z configuration.







Fig. 1. A perspective drawing (*ORTEPII*; Johnson, 1976) of compound (1), with displacement ellipsoids at the 50% level, showing the numbering of the non-H atoms. H atoms are shown as small spheres of arbitrary radii.

Acta Crystallographica Section C ISSN 0108-2701 © 1995 anhydride/sodium acetate. Recrystallization from acetic acid gave pure (1) (m.p. 443-444 K). The propanoic acid derivative used as starting material was obtained by an alkali-catalyzed conversion of *trans*-1,3-bis(3,4dimethoxyphenyl)-2,3-epoxy-1-propanone. The methyl ester, (2), was prepared from (1) and methyl iodide.

A perspective view of (1) is shown in Fig. 1. This crystal structure determination demonstrates unambiguously that acid (1) has the Z configuration.

Crystals of (1) are stabilized by hydrogen bonds. It is notable that the benzylic C(8) atom in (1) is twisted by 0.114 (6) Å out of the plane of the C(11)-C(16)aromatic ring [atom C(7) is located near the C(1)-C(6) plane, with a deviation of 0.023 (8) Å]. The atoms of the carboxyl group are coplanar within 1σ . The plane thus defined forms an angle of $107.7(5)^{\circ}$ with the plane defined by atoms C(1), C(7), C(8), C(11) and C(19). The angle between the aromatic ring planes is $60.1(5)^{\circ}$ in compound (1).

Experimental

Compound (1) was synthesized starting from 2,3-bis(3,4dimethoxyphenyl)-2-hydroxypropanoic acid. This compound was prepared from trans-1,3-bis(3,4-dimethoxyphenyl)-2,3epoxy-1-propanone (Stomberg, Li & Lundquist, 1994) according to a synthetic procedure used for the preparation of substituted mandelic acids (Pandit & Gadre, 1988). A product melting at 434-435 K was obtained (yield 85%). ¹H NMR spectrum (400 MHz, CDCl₃, 293 K, TMS) of 2,3-bis(3,4-dimethoxyphenyl)-2-hydroxypropanoic acid: δ 3.14 (1H, d, J = 13.6 Hz, CH₂), 3.59 (1H, d, J = 13.6 Hz, CH₂), 3.76 (3H, s, OCH₃), 3.84 (3H, s, OCH₃), 3.886 (3H, s, OCH₃), 3.893 (3H, s, OCH₃), 6.5-7.3 p.p.m. (6H, m, H-Ar). For the preparation of (Z)-2,3-bis(3,4-dimethoxyphenyl)propenoic acid, (1), [cf. the method for the preparation of α -arylcinnamic acids described by Gadre & Marathe (1988)], 2,3-bis(3,4-dimethoxyphenyl)-2-hydroxypropanoic acid (3.62 g, 10 mmol) was refluxed with acetic anhydride (100 ml) and fused sodium acetate (2.51 g, 30 mmol) for 1 h. Water (40 ml) was added to the hot reaction mixture, which was then set aside overnight at room temperature. On addition of water (300 ml) a precipitate formed. After 2 h the precipitate was filtered off and dried. The product (3.28 g) consisted primarily of the target molecule, contaminated with the E isomer (thin-layer chromatography, ¹H NMR). A reference sample of the E isomer was prepared according to methods described in the literature (Battersby & Greenock, 1961; Russell & Hunziker, 1969). Two recrystallizations from acetic acid gave 1.98 g of pure Z acid (m.p. 443-444 K; yield 58%). ¹H NMR spectrum (400 MHz, DMSO-d₆, 293 K, TMS): δ 3.76 (3H, s, OCH₃), 3.78 (6H, s, OCH₃), 3.82 (3H, s, OCH₃), 6.9–7.2 [7H, m, H-Ar and vinyl proton (δ 6.94)], 13.3 p.p.m. [1H, s (broad), COOH].

The methyl ester, (2), of (Z)-2,3-bis(3,4-dimethoxyphenyl)propenoic acid was prepared using a procedure analogous to that reported for the preparation of methyl (2,6-dimethoxyphenoxy)acetate (Lundquist, Stomberg & von Unge, 1987). The crude product was freed from tetrabutylammonium iodide by leaching with hot water. Crystallization from ethanol gave a product melting at 421 K (yield 86%). ¹H NMR spectrum (400 MHz, CDCl₃, 293 K, TMS): *δ* 3.82 (3H, *s*, OCH₃), 3.89 (3H, s, OCH₃), 3.902 (3H, s, OCH₃), 3.904 (3H, s, OCH₃), 3.92 (3H, s, OCH₃), 6.8-7.0 p.p.m. (7H, m, H-Ar and vinyl proton).

Crystal data

| $C_{19}H_{20}O_{6}$ | Mo $K\alpha$ radiation |
|---------------------------------|---|
| $M_r = 344.36$ | $\lambda = 0.71073 \text{ Å}$ |
| Orthorhombic | Cell parameters from 25 |
| Pca2 ₁ | reflections |
| a = 8.884(3) Å | $\theta = 11.3 - 19.9^{\circ}$ |
| b = 26.179(2) Å | $\mu = 0.094 \text{ mm}^{-1}$ |
| c = 7.286(2) Å | T = 295 K |
| V = 1694.6 (8) Å ³ | Rod |
| Z = 4 | $0.40 \times 0.12 \times 0.09 \text{ mm}$ |
| $D_x = 1.350 \text{ Mg m}^{-3}$ | Colourless |

Data collection

Rigaku AFC-6R diffractom- $R_{\rm int} = 0.037$ eter $\theta_{\rm max} = 25.0^{\circ}$ $h = 0 \rightarrow 10$ ω scans $k = -26 \rightarrow 31$ Absorption correction: none $l = -7 \rightarrow 8$ 2945 measured reflections 1777 independent reflections 1701 observed reflections $[I > 3\sigma(I)]$

Refinement

| Refinement on F | $(\Delta/\sigma)_{\rm max} = 0.02$ |
|---------------------------------------|--|
| R = 0.034 | $\Delta \rho_{\rm max} = 0.14 \ {\rm e} \ {\rm \AA}^{-3}$ |
| wR = 0.041 | $\Delta \rho_{\rm min} = -0.15 \ {\rm e} \ {\rm \AA}^{-3}$ |
| S = 1.16 | Extinction correction: none |
| 887 reflections | Atomic scattering factors |
| 229 parameters | from International Tables |
| H-atom parameters not | for X-ray Crystallography |
| refined, except for H(O6) | (1974, Vol. IV) |
| $w = 1/[\sigma^2(F_o) + 0.05(F_o)^2]$ | |

3 standard reflections

reflections

monitored every 150

intensity decay: 0.5%

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters ($Å^2$) $= (8 - 2/2) \sum \sum U = 2^{*} 2^$

| | Deq - | $-(6\pi/3)\Delta_i\Delta_j0$ | ıju _i u _j a ι. a _j . | |
|-------|-------------|------------------------------|---|---------|
| | x | у | z | Beu |
| C(1) | 0.3675 (6) | 0.3176 (2) | 0.2183 | 2.9 (2) |
| C(2) | 0.2178 (5) | 0.3337 (2) | 0.230(1) | 3.0 (2) |
| C(3) | 0.1821 (5) | 0.3823 (2) | 0.289(1) | 3.0 (2) |
| C(4) | 0.2949 (6) | 0.4169 (2) | 0.335(1) | 3.2 (3) |
| C(5) | 0.4446 (6) | 0.4016 (2) | 0.319(1) | 3.4 (3) |
| C(6) | 0.4784 (5) | 0.3524 (2) | 0.262(1) | 3.5 (3) |
| C(7) | 0.3989 (6) | 0.2654 (2) | 0.154(1) | 3.3 (3) |
| C(8) | 0.5126 (6) | 0.2346 (2) | 0.200(1) | 2.8 (2) |
| C(9) | -0.0803 (6) | 0.3662 (2) | 0.280(1) | 5.2 (3) |
| C(10) | 0.3603 (7) | 0.4991 (2) | 0.452(1) | 5.2 (3) |
| C(11) | 0.5390 (6) | 0.1835 (2) | 0.117(1) | 3.0 (2) |
| C(12) | 0.4870 (6) | 0.1710 (2) | -0.056 (1) | 2.8 (2) |
| C(13) | 0.5004 (6) | 0.1225 (2) | -0.127 (1) | 3.0 (3) |
| C(14) | 0.5741 (6) | 0.0845 (2) | -0.024 (1) | 3.3 (3) |
| C(15) | 0.6319 (7) | 0.0969 (2) | 0.146(1) | 3.9 (3) |
| C(16) | 0.6141 (6) | 0.1459 (2) | 0.215(1) | 3.9 (3) |
| C(17) | 0.3515 (7) | 0.1386 (2) | -0.392(1) | 4.6 (3) |
| C(18) | 0.6597 (8) | -0.0016 (2) | -0.007(1) | 5.9 (4) |
| C(19) | 0.6268 (7) | 0.2497 (2) | 0.342(1) | 2.7 (2) |
| O(1) | 0.0380 (4) | 0.4013 (1) | 0.302(1) | 4.2 (2) |
| O(2) | 0.2475 (5) | 0.4638 (1) | 0.392(1) | 4.1 (2) |
| O(3) | 0.4521 (4) | 0.1068(1) | -0.295 (1) | 4.5 (2 |

| O(4) | 0.5840 (4) | 0.0378 (1) | -0.107 (1) | 4.2 (2 |
|------|----------------|---------------|------------|--------|
| O(5) | 0.7566 (5) | 0.2596(1) | 0.308(1) | 3.5 (2 |
| O(6) | 0.5722 (4) | 0.2487 (1) | 0.511(1) | 3.5 (2 |
| - | Table 7 Salact | ad accompatri | naramatars | (Å °) |

| Tuble 2. Sele | cica geom | enne paramei | Cr3 (11,) |
|--------------------------------|-------------|--------------|----------------|
| C(1)—C(2) | 1.398 (7) | C(10)—O(2) | 1.432 (7) |
| C(1)—C(6) | 1.379 (7) | C(11)—C(12) | 1.380 (8) |
| C(1)—C(7) | 1.471 (7) | C(11)—C(16) | 1.389 (7) |
| C(2)—C(3) | 1.380(7) | C(12)—C(13) | 1.376 (6) |
| C(3)—C(4) | 1.390 (6) | C(13)—C(14) | 1.409 (7) |
| C(3)O(1) | 1.377 (6) | C(13)—O(3) | 1.360 (7) |
| C(4)—C(5) | 1.394 (7) | C(14)—C(15) | 1.380 (8) |
| C(4)O(2) | 1.362 (6) | C(14)—O(4) | 1.368 (6) |
| C(5)—C(6) | 1.385 (7) | C(15)—C(16) | 1.384 (7) |
| C(7)—C(8) | 1.337 (7) | C(17)—O(3) | 1.412 (7) |
| C(8)—C(11) | 1.486 (7) | C(18)—O(4) | 1.434 (7) |
| C(8)—C(19) | 1.505 (8) | C(19)O(5) | 1.209 (6) |
| C(9)O(1) | 1.405 (7) | C(19)O(6) | 1.321 (6) |
| C(1)—C(7)—C(8) | 128.6 (5) | C(8)-C(19)-C | (6) 113.0 (5) |
| C(2)—C(3)—O(1) | 124.7 (4) | O(5)-C(19)-C | (6) 123.2 (5) |
| C(5)—C(4)—O(2) | 125.3 (5) | C(3)—O(1)—C(| 9) 116.8 (4) |
| C(12)—C(13)—O(3) | 126.1 (5) | C(4)O(2)C(| 10) 117.4 (5) |
| C(15)—C(14)—O(4) | 125.8 (5) | C(13)-O(3)-C | (17) 118.3 (4) |
| C(8)—C(19)—O(5) | 123.8 (5) | C(14)—O(4)—C | (18) 116.5 (5) |
| C(1)—C(7)— | -C(8)-C(11) | _ | 75.9 (5) |
| C(1)-C(7)- | -C(8)-C(19) | | 3.6 (9) |
| C(7)—C(8)— | -C(19)O(5) | - | 110.5 (6) |
| D—H···A | L | DA | D—H···A |
| $O(6) - H(O6) \cdots O(5^{i})$ | 2.0 | 659 (5) | 153 (7) |
| | | | |

Symmetry code: (i) $\frac{3}{3} - x, y, \frac{1}{2} + z$.

The collection of the data was extended to include reflections of the type $h\overline{kl}$. A small number of reflections were affected by a minor contribution from a satellite crystal (usually this was not the case for both a reflection of the type hkl and the corresponding reflection of the type $h\overline{kl}$). The affected reflections were omitted from the data set and average F values were calculated and used in the structure analysis.

Data collection: *TEXSAN* (Molecular Structure Corporation, 1985). Cell refinement: *TEXSAN*. Data reduction: *TEXSAN*. Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1985). Program(s) used to refine structure: *TEXSAN*. Molecular graphics: *TEXSAN*. Software used to prepare material for publication: *TEXSAN*.

Lists of structure factors, anisotropic displacement parameters, Hatom coordinates and complete geometry have been deposited with the IUCr (Reference: CR1190). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Acta Cryst. (1995). C51, 2700-2702

2-[2-(3,4-Dichlorophenyl)vinyl]benzo-[*d*]thiazole

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(Received 1 December 1994; accepted 26 June 1995)

Abstract

The title compound, $C_{15}H_9Cl_2NS$, was synthesized by the condensation of 1-(3,4-dichlorophenyl)-4-methyl-1penten-3-one and 2-aminothiophenol. The title compound is essentially planar in the solid state. The C==C double bond has the *E* configuration.

Comment

The synthesis of the title compound, (III), was unexpected and arose presumably by initial formation of an anil between 1-(3,4-dichlorophenyl)-4-methyl-1-penten-3-one, (I), and 2-aminothiophenol, (II), followed by attack on the carbimino C atom by the S atom leading to the extrusion of the isopropyl group and ring closure forming a benzothiazole. Alternatively, attack by the S atom on the carbonyl C atom of (I) could lead to an intermediate, the amino group of which subsequently forms a bond with this C atom, leading to ring closure and the elimination of the isopropyl group.