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## (Z)-2,3-Bis(3,4-dimethoxyphenyl)propenoic Acid

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#### Abstract

The crystal structure determination demonstrates unambiguously that the title compound, $\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{O}_{6}$, has the Z configuration. The compound was synthesized starting from 2,3-bis(3,4-dimethoxyphenyl)-2-hydroxypropanoic acid, which was in turn obtained by an alkaline treatment of trans-1,3-bis(3,4-dimethoxyphenyl)-2,3-epoxy-1-propanone. ${ }^{1} \mathrm{H}$ NMR data for the title compound and its methyl ester are reported.


## Comment

$\alpha$-Arylcinnamic acids are of interest as intermediates in the syntheses of stilbenes. Gierer, Lenic, Norén \& Szabo-Lin (1974) prepared a series of lignin-related stilbenes using $\alpha$-arylcinnamic acids as starting materials. Such compounds (and their esters) can also be used as starting materials for the preparation of $\alpha$-arylcinnamyl alcohols. Certain compounds of this latter type are intermediates in the acid-catalysed reactions of lignin models of the 1,2-diaryl-1,3-propanediol type (Li \& Lundquist,
1995). The X-ray work presented in this study establishes the structure and stereochemistry of such an $\alpha$-arylcinnamic acid, namely, (Z)-2,3-bis(3,4-dimethoxyphenyl)propenoic acid, (1). Since the synthesis of $\alpha$-arylcinnamyl alcohols from $\alpha$-arylcinnamic acids usually proceeds via the ester derivatives, we also prepared the methyl ester of (1), i.e. (2), and examined it by X-ray crystallographic methods. Attempts to obtain good crystals of (2) failed (the crystals often suddenly cracked, a possible explanation being internal strain). However, a structure analysis was performed and it could be concluded that (2) had the expected $Z$ configuration.

(1)

(2)

The title acid (1), contaminated with minor amounts of the $E$ form, was obtained by heating 2,3 -bis $(3,4-$ dimethoxyphenyl)-2-hydroxypropanoic acid with acetic


Fig. l. A perspective drawing (ORTEPII; Johnson, 1976) of compound (1). with displacement ellipsoids at the $50 \%$ level, showing the numbering of the non- H atoms. H atoms are shown as small spheres of arbitrary radii.
anhydride/sodium acetate. Recrystallization from acetic acid gave pure (1) (m.p. 443-444 K). The propanoic acid derivative used as starting material was obtained by an alkali-catalyzed conversion of trans-1,3-bis(3,4-dimethoxyphenyl)-2,3-epoxy-1-propanone. The methyl ester, (2), was prepared from (1) and methyl iodide.

A perspective view of (1) is shown in Fig. 1. This crystal structure determination demonstrates unambiguously that acid (1) has the $Z$ configuration.

Crystals of (1) are stabilized by hydrogen bonds. It is notable that the benzylic $C(8)$ atom in (1) is twisted by 0.114 (6) $\AA$ out of the plane of the $C(11)-C(16)$ aromatic ring [atom $C(7)$ is located near the $C(1)-C(6)$ plane, with a deviation of 0.023 (8) $\AA$ ]. The atoms of the carboxyl group are coplanar within $1 \sigma$. The plane thus defined forms an angle of 107.7 (5) ${ }^{\circ}$ with the plane defined by atoms $C(1), C(7), C(8), C(11)$ and $C(19)$. The angle between the aromatic ring planes is $60.1(5)^{\circ}$ in compound (1).

## Experimental

Compound (1) was synthesized starting from 2,3-bis(3,4-dimethoxyphenyl)-2-hydroxypropanoic acid. This compound was prepared from trans-1,3-bis(3,4-dimethoxyphenyl)-2,3-epoxy-1-propanone (Stomberg, Li \& Lundquist, 1994) according to a synthetic procedure used for the preparation of substituted mandelic acids (Pandit \& Gadre, 1988). A product melting at $434-435 \mathrm{~K}$ was obtained (yield $85 \%$ ). ${ }^{\prime} \mathrm{H}$ NMR spectrum ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 293 \mathrm{~K}$, TMS) of 2,3-bis(3,4-di-methoxyphenyl)-2-hydroxypropanoic acid: $\delta 3.14(1 \mathrm{H}, d, J=$ $\left.13.6 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 3.59\left(1 \mathrm{H}, d, J=13.6 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 3.76(3 \mathrm{H}$, $\left.s, \mathrm{OCH}_{3}\right), 3.84\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right), 3.886\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right), 3.893$ $\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right), 6.5-7.3$ p.p.m. $(6 \mathrm{H}, m, \mathrm{H}-\mathrm{Ar})$. For the preparation of ( $Z$ )-2,3-bis(3,4-dimethoxyphenyl)propenoic acid, (1), [cf. the method for the preparation of $\alpha$-arylcinnamic acids described by Gadre \& Marathe (1988)], 2,3-bis(3,4-di-methoxyphenyl)-2-hydroxypropanoic acid ( $3.62 \mathrm{~g}, 10 \mathrm{mmol}$ ) was refluxed with acetic anhydride ( 100 ml ) and fused sodium acetate $(2.51 \mathrm{~g}, 30 \mathrm{mmol})$ for 1 h . Water ( 40 ml ) was added to the hot reaction mixture, which was then set aside overnight at room temperature. On addition of water ( 300 ml ) a precipitate formed. After 2 h the precipitate was filtered off and dried. The product $(3.28 \mathrm{~g})$ consisted primarily of the target molecule, contaminated with the $E$ isomer (thin-layer chromatography, ${ }^{1} \mathrm{H}$ NMR). A reference sample of the $E$ isomer was prepared according to methods described in the literature (Battersby \& Greenock, 1961; Russell \& Hunziker, 1969). Two recrystallizations from acetic acid gave 1.98 g of pure $Z$ acid (m.p. $443-444 \mathrm{~K}$; yield $58 \%$ ). ${ }^{1} \mathrm{H}$ NMR spectrum ( 400 MHz , DMSO- $\left.d_{6}, 293 \mathrm{~K}, \mathrm{TMS}\right): \delta 3.76\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right), 3.78(6 \mathrm{H}, s$, $\left.\mathrm{OCH}_{3}\right), 3.82\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right), 6.9-7.2[7 \mathrm{H}, m, \mathrm{H}-\mathrm{Ar}$ and vinyl proton ( $\delta 6.94$ )], 13.3 p.p.m. [ $1 \mathrm{H}, s$ (broad), COOH ].

The methyl ester, (2), of (Z)-2,3-bis(3,4-dimethoxyphenyl)propenoic acid was prepared using a procedure analogous to that reported for the preparation of methyl ( 2,6 -dimethoxyphenoxy)acetate (Lundquist, Stomberg \& von Unge, 1987). The crude product was freed from tetrabutylammonium iodide by leaching with hot water. Crystallization from ethanol gave a product melting at 421 K (yield $86 \%$ ). ${ }^{1} \mathrm{H}$ NMR spectrum
$\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 293 \mathrm{~K}, \mathrm{TMS}\right): \delta 3.82\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right), 3.89$
$\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right), 3.902\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right), 3.904\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right)$, $3.92\left(3 \mathrm{H}, s, \mathrm{OCH}_{3}\right), 6.8-7.0$ p.p.m. (7H, $m, \mathrm{H}-\mathrm{Ar}$ and vinyl proton).

## Crystal data

$\mathrm{C}_{19} \mathrm{H}_{20} \mathrm{O}_{6}$
$M_{r}=344.36$
Orthorhombic
Pca2,
$a=8.884$ (3) $\AA$
$b=26.179(2) \AA$
$c=7.286(2) \AA$
$V=1694.6(8) \AA^{3}$
$Z=4$
$D_{x}=1.350 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\lambda=0.71073 \AA$
Cell parameters from 25 reflections
$\theta=11.3-19.9^{\circ}$
$\mu=0.094 \mathrm{~mm}^{-1}$
$T=295 \mathrm{~K}$
Rod
$0.40 \times 0.12 \times 0.09 \mathrm{~mm}$
Colourless

## Data collection

Rigaku AFC-6R diffractom-

$$
\begin{aligned}
& R_{\text {int }}=0.037 \\
& \theta_{\max }=25.0^{\circ} \\
& h=0 \rightarrow 10 \\
& k=-26 \rightarrow 31 \\
& l=-7 \rightarrow 8
\end{aligned}
$$

3 standard reflections monitored every 150 reflections
intensity decay: 0.5\%

## Refinement

Refinement on $F$
$(\Delta / \sigma)_{\text {max }}=0.02$
$R=0.034$
$w R=0.041$
$S=1.16$
887 reflections
229 parameters
H-atom parameters not
refined, except for $\mathrm{H}(\mathrm{O} 6)$
$\Delta \rho_{\text {max }}=0.14 \mathrm{e}^{-3}$
$\Delta \rho_{\text {min }}=-0.15 \mathrm{e}^{-3}$
Extinction correction: none
Atomic scattering factors from International Tables for X-ray Crystallography (1974, Vol. IV)
$w=1 /\left[\sigma^{2}\left(F_{o}\right)+0.05\left(F_{o}\right)^{2}\right]$
Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters $\left(\AA^{2}\right)$

|  | $B_{\text {eq }}=\left(8 \pi^{2} / 3\right) \sum_{i} \Sigma_{j} U_{i j} a_{i}^{*} a_{j}^{*} \mathbf{a}_{i} \cdot \mathbf{a}_{j}$. |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $B_{\text {eq }}$ |
| C(1) | 0.3675 (6) | 0.3176 (2) | 0.2183 | 2.9 (2) |
| C(2) | 0.2178 (5) | 0.3337 (2) | 0.230 (1) | 3.0 (2) |
| C(3) | 0.1821 (5) | 0.3823 (2) | 0.289 (1) | 3.0 (2) |
| C(4) | 0.2949 (6) | 0.4169 (2) | 0.335 (I) | 3.2 (3) |
| C(5) | 0.4446 (6) | 0.4016 (2) | 0.319 (1) | 3.4 (3) |
| C(6) | 0.4784 (5) | 0.3524 (2) | 0.262 (1) | 3.5 (3) |
| C(7) | 0.3989 (6) | 0.2654 (2) | 0.154 (1) | 3.3 (3) |
| C (8) | 0.5126 (6) | 0.2346 (2) | 0.200 (1) | 2.8 (2) |
| $\mathrm{C}(9)$ | -0.0803 (6) | 0.3662 (2) | 0.280 (1) | 5.2 (3) |
| C (10) | 0.3603 (7) | 0.4991 (2) | 0.452 (1) | 5.2 (3) |
| C(11) | 0.5390 (6) | 0.1835 (2) | 0.117 (1) | 3.0 (2) |
| C(12) | 0.4870 (6) | 0.1710 (2) | -0.056 (1) | 2.8 (2) |
| C(13) | 0.5004 (6) | 0.1225 (2) | -0.127 (1) | 3.0 (3) |
| C (14) | 0.5741 (6) | 0.0845 (2) | -0.024 (1) | 3.3 (3) |
| C(15) | 0.6319 (7) | 0.0969 (2) | 0.146 (1) | 3.9 (3) |
| C(16) | 0.6141 (6) | 0.1459 (2) | 0.215 (1) | 3.9 (3) |
| $\mathrm{C}(17)$ | 0.3515 (7) | 0.1386 (2) | -0.392 (1) | 4.6 (3) |
| C (18) | 0.6597 (8) | -0.0016 (2) | -0.007 (1) | 5.9 (4) |
| C (19) | 0.6268 (7) | 0.2497 (2) | 0.342 (1) | 2.7 (2) |
| $\mathrm{O}(1)$ | 0.0380 (4) | 0.4013 (1) | 0.302 (1) | 4.2 (2) |
| $\mathrm{O}(2)$ | 0.2475 (5) | 0.4638 (1) | 0.392 (1) | 4.1 (2) |
| $\mathrm{O}(3)$ | 0.4521 (4) | 0.1068 (1) | -0.295 (1) | 4.5 (2) |


| $\mathrm{O}(4)$ | $0.5840(4)$ | $0.0378(1)$ | $-0.107(1)$ | $4.2(2)$ |
| :--- | :--- | :--- | ---: | ---: |
| $\mathrm{O}(5)$ | $0.7566(5)$ | $0.2596(1)$ | $0.308(1)$ | $3.5(2)$ |
| $\mathrm{O}(6)$ | $0.5722(4)$ | $0.2487(1)$ | $0.511(1)$ | $3.5(2)$ |

Table 2. Selected geometric parameters $\left(\AA \AA^{\circ}\right)$

| $\mathrm{C}(1)-\mathrm{C}(2)$ | 1.398 (7) | $\mathrm{C}(10)-\mathrm{O}(2)$ | 1.432 (7) |
| :---: | :---: | :---: | :---: |
| $\mathrm{C}(1)-\mathrm{C}(6)$ | 1.379 (7) | $\mathrm{C}(11)-\mathrm{C}(12)$ | 1.380 (8) |
| $\mathrm{C}(1)-\mathrm{C}(7)$ | 1.471 (7) | $\mathrm{C}(11)-\mathrm{C}(16)$ | 1.389 (7) |
| $\mathrm{C}(2)-\mathrm{C}(3)$ | 1.380 (7) | $\mathrm{C}(12)-\mathrm{C}(13)$ | 1.376 (6) |
| $\mathrm{C}(3)-\mathrm{C}(4)$ | 1.390 (6) | $\mathrm{C}(13)-\mathrm{C}(14)$ | 1.409 (7) |
| $\mathrm{C}(3)-\mathrm{O}(1)$ | 1.377 (6) | $\mathrm{C}(13)-\mathrm{O}(3)$ | 1.360 (7) |
| $\mathrm{C}(4)-\mathrm{C}(5)$ | 1.394 (7) | $\mathrm{C}(14)-\mathrm{C}(15)$ | 1.380 (8) |
| $\mathrm{C}(4)-\mathrm{O}(2)$ | 1.362 (6) | $\mathrm{C}(14)-\mathrm{O}(4)$ | 1.368 (6) |
| $\mathrm{C}(5)-\mathrm{C}(6)$ | 1.385 (7) | $\mathrm{C}(15)-\mathrm{C}(16)$ | 1.384 (7) |
| $\mathrm{C}(7)-\mathrm{C}(8)$ | 1.337 (7) | $\mathrm{C}(17)-\mathrm{O}(3)$ | 1.412 (7) |
| $\mathrm{C}(8)-\mathrm{C}(11)$ | 1.486 (7) | $\mathrm{C}(18)-\mathrm{O}(4)$ | 1.434 (7) |
| $\mathrm{C}(8)-\mathrm{C}(19)$ | 1.505 (8) | $\mathrm{C}(19)-\mathrm{O}(5)$ | 1.209 (6) |
| $\mathrm{C}(9)-\mathrm{O}(1)$ | 1.405 (7) | $\mathrm{C}(19)-\mathrm{O}(6)$ | 1.321 (6) |
| $\mathrm{C}(1)-\mathrm{C}(7)-\mathrm{C}(8)$ | 128.6 (5) | $\mathrm{C}(8)-\mathrm{C}(19)-\mathrm{O}(6)$ | 113.0 (5) |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{O}(1)$ | 124.7 (4) | $\mathrm{O}(5)-\mathrm{C}(19)-\mathrm{O}(6)$ | 123.2 (5) |
| $\mathrm{C}(5)-\mathrm{C}(4)-\mathrm{O}(2)$ | 125.3 (5) | $\mathrm{C}(3)-\mathrm{O}(1)-\mathrm{C}(9)$ | 116.8 (4) |
| $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{O}(3)$ | 126.1 (5) | $\mathrm{C}(4)-\mathrm{O}(2)-\mathrm{C}(10)$ | 117.4 (5) |
| $\mathrm{C}(15)-\mathrm{C}(14)-\mathrm{O}(4)$ | 125.8 (5) | $\mathrm{C}(13)-\mathrm{O}(3)-\mathrm{C}(17)$ | 118.3 (4) |
| $\mathrm{C}(8)-\mathrm{C}(19)-\mathrm{O}(5)$ | 123.8 (5) | $\mathrm{C}(14)-\mathrm{O}(4)-\mathrm{C}(18)$ | 116.5 (5) |
| $\mathrm{C}(1)-\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(11)$ |  | -175.9 (5) |  |
| $\mathrm{C}(1)-\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(19)$ |  | 3.6 (9) |  |
| $\mathrm{C}(7)-\mathrm{C}(8)-\mathrm{C}(19)-\mathrm{O}(5)$ |  | -110.5 (6) |  |
| $D-\mathrm{H} \cdots \cdot$ |  | $D \cdots A$ | D-H. . A |
| $\mathrm{O}(6)-\mathrm{H}(\mathrm{O} 6) \cdots \mathrm{O}\left(5^{\mathrm{i}}\right)$ |  | 2.659 (5) 153 |  |
| Symmetry code: (i) $\frac{3}{2}-x, y, \frac{1}{2}+z$. |  |  |  |

The collection of the data was extended to include reflections of the type $h \bar{k} \bar{l}$. A small number of reflections were affected by a minor contribution from a satellite crystal (usually this was not the case for both a reflection of the type $h k l$ and the corresponding reflection of the type $h \bar{k} \bar{l}$ ). The affected reflections were omitted from the data set and average $F$ values were calculated and used in the structure analysis.
Data collection: TEXSAN (Molecular Structure Corporation, 1985). Cell refinement: TEXSAN. Data reduction: TEXSAN. Program(s) used to solve structure: SHELXS86 (Sheldrick, 1985). Program(s) used to refine structure: TEXSAN. Molecular graphics: TEXSAN. Software used to prepare material for publication: TEXSAN.

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## 2-[2-(3,4-Dichlorophenyl)vinyl]benzo[d]thiazole

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#### Abstract

The title compound, $\mathrm{C}_{15} \mathrm{H}_{9} \mathrm{Cl}_{2} \mathrm{NS}$, was synthesized by the condensation of 1-(3,4-dichlorophenyl)-4-methyl-1-penten-3-one and 2 -aminothiophenol. The title compound is essentially planar in the solid state. The $\mathrm{C}=\mathrm{C}$ double bond has the $E$ configuration.


## Comment

The synthesis of the title compound, (III), was unexpected and arose presumably by initial formation of an anil between 1-(3,4-dichlorophenyl)-4-methyl-1-penten-3-one, (I), and 2-aminothiophenol, (II), followed by attack on the carbimino $C$ atom by the $S$ atom leading to the extrusion of the isopropyl group and ring closure forming a benzothiazole. Alternatively, attack by the $S$ atom on the carbonyl $C$ atom of (I) could lead to an intermediate, the amino group of which subsequently forms a bond with this $C$ atom, leading to ring closure and the elimination of the isopropyl group.


[^0]:    Lists of structure factors, anisotropic displacement parameters, H atom coordinates and complete geometry have been deposited with the IUCr (Reference: CR1190). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England.

